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Source: Purcell + Kotz, Inorganic Chemistry, 1977 These are determined by quantum numbers l and m_l . Resulting orbital shapes: Note that the orbitals are 3D waves! All atomic orbitals of a given atom must be orthogonal to one another. The surface is typically the 75% probability boundary.

1.3 Review of atomic orbitals and their shapes

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The elements Name Symbol Atomic Molar mass Name Symbol

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Atomic Molar mass number (g mol⁻¹) number (g mol⁻¹)
Actinium Ac 89 227 Meitnerium Mt 109 268 Aluminium
(aluminum) Al 13 26.98 Mendeleevium Md 101 258 Americium Am
95 243 Mercury Hg 80 200.59 Antimony Sb 51 121.76
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6.0 Main Group Elements: Typical Chemistry and Compounds 6 ...

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Chemistry of the Elements I: Main group chemistry

Five mononuclear photoluminescent zinc(II) complexes have been synthesized from the tetradentate ligand L,

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N,N'-bis(2-pyridylmethylene)-1,3-propanediamine, and characterized by various spectroscopic techniques, elemental analysis and X-ray crystallography. The geometry around the Zn²⁺ centre is distorted square pyramidal, although a significant distortion towards the trigonal pyramidal ...

Five new mononuclear zinc(II) complexes with a ...

The effects of inorganic salts on the binding of cis-diamminedichloroplatinum(II) (= cis-Pt) to salmon sperm DNA have been investigated for various concentrations of NaCl, NaHCO₃, and NH₄HCO₃. In concentrations of 0.01 M or higher, all three salts strongly inhibit the binding to DNA. However, only NH₄HCO₃, which may generate NH₃, is able to block irreversibly the reactive coordination ...

A simple method for the inactivation of monofunctionally

...

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The general VB solution for the series of EX n molecules and examples are give in the following tables: Source for all tables: Purcell + Kotz, Inorganic Chemistry, 1977) Huheey, Keiter, Keiter, Inorg. Chem. 4th Ed., 1993.

4. Valence Bond (VB) Theory - University of Guelph

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inorganic and organo- metallic complexes, it clearly appears that out of 26 cases covering both high spin and low spin situations, 21 represent paramagnetic species (K. Purcell, J. Kotz, Inorganic Chemistry , Saunders, 1977, p561). This would suggest that, if there is a part in chemistry to illustrate the

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